

Volume 1 | Issue 1

Article 4

Effects of Anisotropy on the Hydrogen Induced Fatigue Crack Propagation of a Banded Ferrite/Pearlite Steel

L. Tau

Institute of Materials Science and Engineering National Taiwan University Taipei Taiwan 10764 R.O.C.

S.L.I. Chan

Institute of Materials Science and Engineering National Taiwan University Taipei Taiwan 10764 R.O.C.

C.S. Shint

Department of Mechanical Engineering National Taiwan University Taipei Taiwan 10764 R.O.C

Follow this and additional works at: https://jmstt.ntou.edu.tw/journal

Part of the Engineering Commons

Recommended Citation

Tau, L.; Chan, S.L.I.; and Shint, C.S. (1993) "Effects of Anisotropy on the Hydrogen Induced Fatigue Crack Propagation of a Banded Ferrite/Pearlite Steel," *Journal of Marine Science and Technology*. Vol. 1: Iss. 1, Article 4. DOI: 10.51400/2709-6998.2474

Available at: https://jmstt.ntou.edu.tw/journal/vol1/iss1/4

This Research Article is brought to you for free and open access by Journal of Marine Science and Technology. It has been accepted for inclusion in Journal of Marine Science and Technology by an authorized editor of Journal of Marine Science and Technology.

Effects of Anisotropy on the Hydrogen Induced Fatigue Crack Propagation of a Banded Ferrite/Pearlite Steel

Acknowledgements

The authors would like to appreciatively ac-knowledge the support of this work by the NationalScience Council of Republic of China under Contract no. NSC 79-0405-E-002-18.

This research article is available in Journal of Marine Science and Technology: https://jmstt.ntou.edu.tw/journal/vol1/ iss1/4

EFFECTS OF ANISOTROPY ON THE HYDROGEN INDUCED FATIGUE CRACK PROPAGATION OF A BANDED FERRITE/PEARLITE STEEL

L. Tau*, S.L.I. Chan* and C.S. Shin⁺

 Institute of Materials Science and Engineering National Taiwan University Taipei Taiwan 10764 R.O.C.
[†]Department of Mechanical Engineering National Taiwan University mTaipei Taiwan 10764 R.O.C.

Key words: fatigue crack propagation, hydrogen embrittlement, banded structure, inclusion.

ABSTRACT

The fatigue crack propagation (FCP) properties of AISI 4130 steel plates with a ferrite/pearlite (F/P) banded structure were evaluated both in air and electrolytic hydrogen charging environments. To understand the combined effects of hydrogen, directions of banding and elongated non-metallic inclusions on the FCP, specimens with six different orientations relative to the rolling direction were sampled and charged with hydrogen. When tested in air, the result reveals that the crack plane as related to the direction of banding, and the local stress concentration induced by elongated inclusions, largely affected the FCP rate. It also shows that the relative dominance of banding and inclusions depended on the direction of banding. For specimens after hydrogen charging, the results show the FCP rate was an order higher than that for the specimens tested in air. The enhancement of FCP rate was found to vary with respect to the main three directions (longitudinal, transverse and through-surface directions) as related to banding. For specimens where hydrogen can easily diffuse along the longitudinal direction into specimens and assisted crack propagation, they had the highest FCP rate. It was followed by the specimens in which hydrogen diffused along transverse direction. When hydrogen diffused through layers of ferrite

INTRODUCTION

For Cr-Mo and Ni-Cr-Mo steel plates after hot rolling or controlled rolling, two types of preferred orientation can occur, namely the crystallographic texturing (alignment of crystallographic planes and directions) and mechanical fibering (alignment of grains, chemically segregated areas and inclusions). Thus a typical microstructure of hot rolled Cr-Mo and Ni-Cr-Mo steel plates usually composed of a structure with alternative layers of banded ferrite and pearlite (F/P) structure [1]. The anisotropy of banded structure due to mechanical fibering is generally manifested by a decrease in tensile ductility, impact resistance and fracture toughness in the long and short transverse directions [2]. In addition, it has been found that at room temperature the main diffusive path for hydrogen in ferrite/pearlite structure is along the ferrite grains or interface between ferrite and pearlite [3]. This means that the hydrogen intake is different in

Table 1. The chemical composition of AISI 4130 steel.

Element	C	Si	Mn	Р	S	Ni	Cr	Мо	Fe
wt.%	0.3	0.2	0.41	0.02	0.01	0.04	0.99	0.15	Bal

different directions when banding occurs [4]. Thus the degree of hydrogen embrittlement, as manifested by the loss of ductility, is similar when hydrogen diffuses along the longitudinal and transverse directions. But if the banding is perpendicular to the hydrogen diffusion path, the degree of hydrogen embrittlement is reduced. Above observations make the anisotropic effects of the banding on FCP in air and hydrogen environment interesting [5,6]. On the other hand, elongated sulfide inclusions have been noted to assist short range hydrogen diffusion [7,8] and lead to differences in behavior in hydrogen induced cracking in different orientations [9,10]. It has also been stated that the dissolving of manganese sulfide inclusions in an aqueous hydrogen environment leads to local concentration of H_2S , which can accelerate hydrogen embrittlement [7,11].

In the above investigations the degree of embrittlement was studied under static stress or slow strain rate conditions. The role of banding on the hydrogen induced fatigue cracking has scarcely been mentioned. It was the objective of this work to study the effects of the F/P striations and elongated inclusions in AISI 4130 steel plates on the FCP rate in a hydrogen charging environment.

EXPERIMENTAL PROCEDURES

1. Materials and specimen sampling

The chemical composition of the 32 mm thick AISI 4130 plates used in this study is given in Table 1. The microstructure of as-received material was a typical banded ferrite/pearlite structure (as shown in Fig. 1). The inclusions in banded structure were elon-



Fig. 1. Micrograph showing the banded ferrite/pearlite structure.

gated along the rolling direction (as shown in Fig. 2).

Fatiguecrack propagation tests were performed using the appropriate ASTM specifications[12]. Fatigue specimens of compact-tension type were machined from all six possible orientations: LS, LT, TS, TL, ST and SL orientations as related to the rolling direction (as shown in Fig. 3). The dimensions of these fatigue specimens are: width = 25mm, thickness = 6.25mm.

2. Fatigue Crack Propagation Rate Test

To study the FCP properties of the AISI 4130 steel plates under hydrogen charging, the specimens



rolling direction





Fig. 3. Schematic diagram showing the sampling orientations of the tested specimens.

were charged in 4Vol.% $H_2SO_4 + 20\text{mg } As_2O_3/1$ solutions, with an applied cathodic current density 20A/m². Other conditions of loading were: *R*-ratio = 0.1, P_{max} = 3.5 kNt and loading frequency = 0.5Hz. Fatigue test was also performed in air, with a loading frequency of 20Hz. The FCP data in air have been summarized using the Paris' law, and plots of the FCP data in air were used to contrast with that in the

RESULTS AND DISCUSSION

1. FCP rate in air

cathodic charged conditions.

The FCP data of all specimens in air were treated by Paris' law, and the constants are rearranged as in Table 2. As discussed in reference 5, the crack paths of specimens with a banded structure can be divided into three modes. The first mode is for the specimens of SL and ST orientation, fatigue crack propagated easily along weak interfaces, thus delamination and a higher FCP rate would resulted. The specimens of TL and LT orientation have a second mode, in which the path of crack was divided by alternate layers of ferrite and pearlite, and this composite-like effect was found to resist FCP rate. The third mode is found in the specimens of TS and LS orientation, where the crack propagation through direction of thickness would branch easily, and the FCP rate would be lowered. In this work, for the FCP the m value in the Paris' law agreed closely to this viewpoint, except for the specimens of TL orientation. The reason for this apparent discrepancy is that the influences on FCP of specimens with a banded structure include both the banding and elongated inclusions. The result reveals that the local stress concentration induced by elongated inclusions, which are parallel to the crack plane, produced a pronounced effect on FCP rate for the specimens of TL orientation. Similar observation is found for the SL specimens, which have a higher m value as compared with that of SL specimens.

Table 2. The parameter C, m in Paris' law : da/dn = C (K) ^m, for FCP in air, 20Hz.

orientation		C(mm/cycle, MPa-m ^{1/2})	m	
banded	TL	7.34*10-10	3.62	
structure	TS	1.68*10-10	3.27	
	LT	1.81*10-9	3.25	
	LS	2.03*10-9	3.28	
	SL	4.49*10-10	3.83	
	ST	6.41*10-10	3.66	

2. FCP rate in hydrogen environment

After hydrogen charging, the FCP rates of all specimens were an order higher than that of the specimens tested in air, regardless of the orientations. The FCP rates in the two environments as represented by a graph of log da/dn versus log Δk , as shown in Fig. 4. It also shows that the variation of FCP rates of different orientations were more significant when hydrogen charged. This was especially so in the zone of higher Δk .

For clarity curves of log da/dn versus log k for specimens charged with hydrogen are shown in Fig. 5.



Fig. 4. log da/dn versus log ∆K curve for specimens tested in air and hydrogen environments.



Fig. 5. log da/dn versus log ∆K curve for banded structure in hydrogen environment.

The result shows that the FCP rates of specimens with a banded structure could be divided by three conditions according to their degree of enhancement. Figure 5 shows that specimens of ST and TS orientations had the highest FCP rate, followed by the specimens of LS and SL orientations, with the specimens of TL and LT orientations having the lowest crack propagation rate. As discussed in reference 4, Lee and Chan indicated that the hydrogen permeabilities were different for banded structure. The penetration flux of hydrogen along the longitudinal direction was higher than that along the transverse direction (as shown in Fig. 6), which in turn was faster than that along the through-face direction. Above results suggest that the preferred path for hydrogen diffusion in a banded structure has an important effect on fatigue crack propagation. For the ST and TS specimens, the faces where hydrogen diffused in have the ferrite/pearlite bands parallel to the hydrogen diffusion path, i.e. along the longitudinal direction. It means that the hydrogen intake was the highest [4] and there was an ample supply of hydrogen to the crack plane. Consequently the FCP rate for the specimens was the highest. On the other hand, for the T-L and L-T specimens, the hydrogen had to diffuse through bands of ferrite and pearlite, which was obviously more difficult. Thus the FCP rate was the lowest. It has also been found that non-metallic inclusions did not affected very much the FCP rate. As is demonstrated by the S-L and T-L specimens, both specimens only have a moderate FCP rate compared with the others.

CONCLUSION

- 1. In air, the alignment of ferrite/pearlite and elongated inclusion both affect the FCP rates of specimens with a banded structure. It was found that the *SL*, *ST*specimens have the highest FCP rates. Inclusions in the *SL* and *TL* specimens were parallel to the crack plane, hence further assisted the FCP rate.
- 2. In hydrogen environment, the FCP rates of all specimens was an order higher than that when tested in air. The FCP rate was very much depended on the hydrogen diffusion path to the crack tip. When thecrack plane were parallel to the longitu dinal fast diffusion path, the FCP rate was the highest. When the crack plane were perpendicular to the high hydrogen diffusivity path, the FCP rate was found to be lowered.

ACKNOWLEDGMENT

The authors would like to appreciatively acknowledge the support of this work by the National Science Council of Republic of China under Contract no. NSC 79-0405-E-002-18.





REFERENCE

- 1. Lavender, J.D. and F.W. Jones, J. Iron Steel Inst., London, 9 (1949) 14.
- 2. Grange R.A., Metall. Trans., 2 (1971) 417.
- 3. Chan S.L.I. and J.A. Charles, Mater. Sci. Technol., 2 (1986) 956.
- 4. Lee H.L. and S.L.I. Chan, Mater. Sci. and Eng., A170 (1991) 193.
- 5. Heiser F.A. and R.W. Hertzberg, J. of Basic Eng., June 1971, p.211.
- 6. Wilson A.D., J. of Engineering Materials and Technology, July 1984, p.233.
- 7. Pumphrey P. H., Corrosion-NACE, 36 (1980) 537.
- 8. Pumphrey P.H., Metal Science, 16 (1982) 41.
- 9. Savage W.F., E.F. Nippes and M.A. Pellman, Welding J., Nov. 1979, p.315.
- 10. Ven Katasubramanian, T.V. and T.J. Baker, Metal Science, 16 (1982) 543.
- 11. Vosikovsky O., J. of Engineering Materials and Technology, Oct. 1975, p.298.
- 12. 1982 Annual Book of ASTM Standards, E647, ASTM, Philadelphia, 1982.