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# CHARACTERISTIC STUDY OF MHD MINIPUMP

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Key words: MHD minipump.

#### ABSTRACT

The present study aims to develop and discuss the availability of the Magneto Hydrodynamics Minipump (MHD) as a miniaturizative part specified for precise manufacturing or medical application. Based on the MHD theory, an available working current (1.8A) induced by lower voltage (0-12V) and weaker magnetic intensity (0.04T) meets the need in mini-system operation. However, precise estimations of working capacity, maximum head, dynamic head, and static head in various electrolyte concentrations are necessary. Furthermore, this work also provides a simplified analytic model with a static balance between Lorentz force, surface tension force, and body force. Compared with analytic solutions and experimental results, the static head depends on the operating voltage, and the higher dynamics head is produced instantly in the working medium of NaCl • H<sub>2</sub>O which is more appropriate for the impulse pump.

#### I. INTRODUCTION

Magneto hydrodynamics (MHD), which primarily describes the flow dynamics by coupling the interaction of electrically conducting current and electro-magnetic intensity, has gradually drawn public attention to the field of micro fluids. Though the basic MHD principle began to be discussed in the last decade, the effective application in various fields has been studied only recently. Geng, X., Yuan, H. and Oguz, H.N. [1] calculated the size of bubbles in the pumping process and discussed the availability of the MHD pump in the micro channel. Huang et al. [2] also demonstrated the pumping characteristic with several aqueous solutions and found that the formation of bubbles at the electrode significantly affected the performance of the pump. Jang and Lee [3] published a paper on the MHD minipump and evaluated its performance by measuring the pressure head and regulating the voltage range from 10-60V under a magnetic flux density of 0.19-0.44T. Luis et al. [4] proposed a finite element method to simulate the MHD duct flow with a formulation of a

penalty function. Extended analysis proved the Hartman number to be a key parameter which significantly influenced the flowing behavior of the electrical conducting liquid. The simulation results showed that the restricted Hartman number  $10^2 \sim 10^3$  would achieve a stable solution which gave a reasonable flow distribution. Wang *et al.* [5] provided a simplified 2-D MHD model assuming a steady state with an incompressible and fully developed flow. Under this condition, the Lorentz force was converted into pressure terms and occurring frictional effects on the channel sidewalls were also taken into account.

This research sets up an experimental model with two electrolytes in different concentrations. The designed minipump with a distance of 3 millimeter between the electrodes operates in smaller input voltage (may be in series with the usage of 3 V batteries); which is quite different from the operation in higher voltages or stronger magnetic fields in previous experiments. Moreover, an analytic solution with static head is also derived to identify the experiment results in this study.

#### **II. FUNDAMENTAL ANALYSIS**

The experimental apparatus is sketched in Figs. 1~ 3, which consists of four major systems: power supply system, hydraulic head measuring tube, image recording system and test channel. A power supply (Gpc-3030) with a voltage range of  $0V \sim 30V$ , and a current between  $0A \sim 3A$  was used. Copper sheets with a physical dimension of  $60 \times 3 \times 0.2$  mm<sup>3</sup> placed on the top and bottom of a test channel made up by thin acrylic plates were used as electrodes, and then the channel sidewalls were equipped with high-density magnetism with a magnetic flux of 0.04 T (Fig. 2). To measure the static head of electrolyte, an acrylic measuring tube with a diameter of 1 mm and an inclined angle of 30 degree was sealed at the right side (Fig. 3). A camcorder (DCR-HC 20) connected to a personal computer through a traducer wire captured the recorded image and transmitted it to the personal computer in real time.

Initially, the test channel was filled with electrolyte and surface tension force elevated the liquid to height  $h_0$ . This is defined as the equilibrium between surface tension and elevated liquid weight.

$$\sigma \ 2 \pi r \sin \theta = \rho \pi \ r^2 gh_0 \tag{1}$$

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Fig. 1. Experimental apparatus (1. Power source supply; 2. Test channel; 3. CuSO<sub>4</sub> • 5H<sub>2</sub>O, NaCl; 4. Halogen lamp; 5. Camera DCR HC-20; 6. Personal computer).



Fig. 2. Photograph of test channel.



Fig. 3. Cross section view of test channel.

When the external voltage induced the parallel working current, the electrolyzing process started simultaneously. Once the induced current interacted with the normal magnetic field, the pumping motion began. This motion immediately raised the working medium until the final balance between body force, surface tension force, and Lorentz force was reached.

$$J \times B \times L + \sigma 2\pi r \sin \theta = \rho \pi r^2 (h_0 + \Delta h)g$$

Substituting (1) into above equation, the final expression could be simplified as:

$$\Rightarrow J \times B \times L = \rho \pi r^2 \Delta hg \tag{2}$$



Fig. 4. Bubble's generation for NaCl • H<sub>2</sub>O at t=5s under an applied voltage of 9V.



Fig. 5. Bubble's generation for CuSO<sub>4</sub> • 5H<sub>2</sub>O at t=5s under an applied voltage of 9V.

#### **III. RESULT AND DISCUSSION**

#### 1. Induced Current and Maximum Head with Time

The significant difference between the two aqueous solutions, NaCl aqueous solution and  $CuSO_4 \cdot 5H_2O$  aqueous solution, used in this study is the amount of bubbles.

Figure 4 shows that NaCl solution easily produces massive bubbles (Fig. 4), and then  $CuSO_4 \cdot 5H_2O$  produces less under a voltage of 9V, and magnetic flux of 0.04T (Fig. 5). Due to the electrolyte with higher electric conductivity (100 ms/cm) in 5% NaCl solution, the induced current jumped up to 1.8A immediately within the first second (see Fig. 6) resulting in the formation of a massive amount of tiny bubbles in the channel, which reduces the electrical conductivity in the working medium, and made the induced current drop from 1.8 A to 0.25 A within t=1s ~ t=5s (Fig. 6). In contrast to the NaCl solution, the 5%,  $CuSO_4 \cdot 5H_2O$  solution with lower electrical conductivity (30 ms/cm) induced less working current which slowed the growth of the bubbles to t=5s (Fig. 5). Here, a steady current of 0.8A (Fig. 6) and a final static head of 5mm (Fig. 7) were observed.

Compared to Figs. 6 and 7, similar distributions are shown in maximum head and working current with time. For the NaCl solution, the head rapidly climbs to a peak value of 14mm in the first second (Fig. 7). This rise is analogous to the current distribution in Fig. 6, where the maximum current approximately reaches 1.8A. Afterwards, the hydraulic head lowers to 2mm instantaneously and then an unstable fluctuation between 2mm



Fig. 6. Variation of induced current and time under an applied voltage of 9V.



~ 3mm occurred before the final static head reaches in the next 2 seconds (Fig. 7). This result corresponds to the decline of the induced current in Fig. 6 where the weaker Lorentz force is no longer able to support the liquid medium, so the hydraulic head drops down immediately with a sharp fluctuation (Fig. 7). As for the discussion in  $CuSO_4 \cdot 5H_2O$  solution, the hydraulic head in Fig. 7 gradually increases to a maximum value of 6mm initially and shows a slow decrease with a slight fluctuation in five seconds. This result also agrees with the current distribution in Fig. 6, where the induced current slowly rises to 0.8 A and then remains constant.

## 2. Maximum Head, Analytic Head, and Dynamic Head with the Induced Current

In dealing with the maximum head and induced current in NaCl solution, Fig. 8 shows that the maximum head increases linearly as the current varies (0.4 A, 4mm ~ 1.8 A, 15mm). The analytic head with its final static state also presents a linear relationship with the induced current (0.4 A, 2mm ~ 1.8A, 9mm). The discrepancy between both distributions can be defined as the dynamic head caused by initial fluid acceleration with active Lorentz force. Therefore kinetic energy in the dynamic head can be determined by the working current and faster flow velocity. As for CuSO<sub>4</sub> · 5H<sub>2</sub>O solution in Fig. 9, both distributions of the maximum head (0.2 A, 2mm ~ 0.8A, 6mm) and the analytic head (0.2 A, 1.7 mm ~ 0.8 A, 4mm) are so close that the difference between them is insignificant, yet both will slightly increase with an induced current. This leads to the



Fig. 8. Comparison between maximum head, static head, and analytic head with maximum current for NaC1 • H<sub>2</sub>O.



prediction that the dynamic head of the lower conducting medium,  $CuSO_4 \cdot 5H_2O$ , is small and nearly independent of the induced current in the pumping process. Therefore the liquid will be slowly and steadily elevated, which is dissimilar to the result in higher electrically conductivity solution (NaCl).

Figs. 8 and 9 also indicate that the analytic solutions in equation (2) are well in agreement with the experimental static heads. That implies an easier option by using the simpler equation to determine the static head and its availability in this study.

#### 3. Maximum Head with Aqueous Concentration in Different Voltage Applied

The maximum head distribution of NaCl solution with a concentration of 1% and 5% is discussed in Fig. 10. Though both distributions nearly keep a linear relation with the applied voltage, the maximum heads of 0mm ~ 2mm in a 1% concentration are much less than those of 0mm ~ 16mm in a 5% concentration and their difference extends as voltage increases (0 mm, 1V ~ 13 mm, 9V). A denser solution with higher electrical conductivity produces stronger instantaneous Lorentz force to raise the working medium and makes a result of significant deviation for the maximum head in both cases; especially under the condition of higher applied voltage. In contrast to the above discussion, both maximum heads in 1% and 5% CuSO4 • 5H2O solutions slightly vary with the increase of applied voltage (Fig. 11). Their difference, 0mm~3mm, is relative small and nearly insensitive to variation of concentrations. This smaller divergence tells that the weak MHD effect will be not much improved



Fig. 10. Comparison between the maximum head of 5% NaCl and 1% NaCl solutions with an applied voltage of 3V-11V.



Fig. 11. Comparison between the maximum head of 1% and 5%CuSO<sub>4</sub> • 5H<sub>2</sub>O solutions with an applied voltage of 3V-11V.

with lower electrical conductivity solution of CuSO4  $\cdot$  5H2O, even in the usage of higher concentrations.

#### **IV. CONCLUSION**

Several important results from above discussion are concluded as follows:

- 1. Bubbles produced in higher induced current not only retard flow motion, but also lower the hydraulic head.
- 2. Due to the required low flow rate in the minipump, lower input voltage with weaker magnetic intensity is adequate to

produce Lorentz force to overcome the flow drag and reduce the birth of bubbles.

- 3. Maximum heads with a time in higher electrical conductivity solution are similar to the distribution of delta function and suitable to be applied in an impulse pump. Nevertheless, the stable growth of the maximum heads in a lower electrical conductivity solution fits the continuous type pump.
- 4. With the expansion of applied voltage, the height of both maximum and dynamic heads will be significantly enhanced in higher electrical conductivity solution.
- 5. A simplified steady solution in an analytic model is of special advantage to determine the static head which has been proven to be in excellent agreement with the experimental data.

#### NOMENCLATURE

- $\sigma$  : surface tension coefficient
- $\rho$  : density of working medium
- r : radius of static tube
- $h_0$  : surface tension head
- J : induced current density
- *B* : magnetic flux density
- L : the length of the test channel
- $\Delta h$  : Lorentz static head

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