Structure and photocatalytic influence of N-doping ATO nano-tubes on antibacterial activity toward Escherichia coli

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RESEARCH ARTICLE

Structure and Photocatalytic Influence of N-doping ATO Nano-tubes on Antibacterial Activity Toward *Escherichia coli*

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Abstract

The photocatalytic antibacterial properties of anodic titanium oxide (ATO) nanotubes in the anatase structure derived from annealing at 450 °C or 550 °C or N-doping treatment were tested on *Escherichia coli* under UV-A, visible light illumination, and dark conditions. Under UV-A illumination and dark conditions, in most cases, *E. coli* cultures containing ATO nanotubes of different treatments showed either no or slight differences in their growth from the cultures without ATO nanotubes. However, under visible light illumination, the growth of *E. coli* was inhibited in cultures containing ATO nanotubes. Among the three forms of ATO nanotubes, N-doped ATO nanotubes had the strongest antibacterial activity. While the antibacterial activities of both the annealed and the N-doped ATO nanotubes increased positively correlated to the increase in their total surface area.

Keywords: Visible-light, Photocatalytic antibacterial activities, Anatase TiO₂, N-doping, UV-A, *Escherichia coli*

1. Introduction

A semiconductor of titanium dioxide (TiO₂) has an outstanding photocatalytic feature. Harmful organic and unwanted compounds can be destroyed by its strong oxidizing power. More importantly, TiO₂ has been extensively applied to microorganisms in contaminated water and to air treatment [1,2]. The current research status of the photocatalytic effect of TiO₂ on antibacterial application was reviewed in the past five years. Transparent coatings produced by stirring a TiO₂-bearing suspension showed significant antibacterial activity toward *Escherichia coli* under low UV irradiation [3]. The photocatalytic antibacterial effects of anatase have been investigated under different irradiation methods. It has been reported that irradiation of

wavelengths >390 nm produced negligible photocatalytic bactericidal results on the surface of anatase. Wavelengths of UV-A/VIS light in the range of 380–390 nm were found to have a photocatalytic bactericidal effect similar to that of UV-A [4]. A comparison of the synergetic effects of PoPD (poly(o-phenylenediamine)), TiO₂@PoPD (titania@poly(o-phenylenediamine)) and TiO₂ was based on slow charge recombination and rapid charge separation. The photocatalytic activities of TiO₂@PoPD photocatalysts were remarkably enhanced in solar light. In addition, the antibacterial activity of TiO₂@PoPD core-shell nanocomposite has a greater effect against different pathogenic bacteria, such as Gram negative (*Klebsiella pneumonia*) and Gram positive (*Bacillus subtilis*) bacteria than do PoPD and TiO₂ in a diffusion process [5]. The

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antimicrobial properties of Ag–TiO₂–PVDF (polyvinylidene fluoride) ultrafiltration membranes can effectively prevent biofilm formation and bacteria on the surfaces of the membranes. Water treatment often involves the use of Ag–TiO₂–PVDF membranes [6]. Highly photoactive thin films of Cu doped anatase TiO₂ were deposited on glass substrates by aerosol-assisted chemical vapor deposition (AACVD) [7]. The films exhibited strongly-enhanced photocatalytic activity relative to that of pure anatase and excellent antibacterial ability (vs. *E. coli* and *Staphylococcus aureus*) [7]. The exciton lifetimes were extended in Cu doped anatase, and the enhancement of the UV photoactivity can be explained by the interactions inside the anatase lattice between substitutional and interstitial Cu [7].

Reactive oxygen species (ROS), namely, hydroxyl •OH radicals and superoxide anion •O₂[−] radicals, can be produced by TiO₂ exposed to light and have an antibacterial photocatalytic effect. These ROS have the oxidizing power to, for example, decompose various organic compounds in the outer membranes of bacteria into harmless CO₂ and H₂O. The strong oxidizing ability of TiO₂ is derived from the energy gap of the photo-induced electrons in the conduction band (e[−] cb), •O₂[−] radicals formed by reducing O₂, positive holes in the valence band (h⁺ vb), and •OH radicals formed by the reaction with H₂O upon ultraviolet light (UV, λ < 380 nm) excitation [8–10]. UV light is harmful to human beings, and this fact restricts the application of TiO₂ in industry. Extensive investigations of the improvement in the photocatalytic antibacterial activity of TiO₂ have been conducted with approaches such as fabricating TiO₂ into nanotube layers to increase the surface area and doping it with impurities, e.g., other elements (C, N, F, P and S) or transition metals. In a previous work, it was shown that a TiO₂ nanotube layer with an amorphous structure can be changed to the anatase form (anatase TiO₂; ATO) by annealing in air at 450 °C. Considerable success has also been achieved by reducing the band gap energy of ATO by doping [11]. N-doping of ATO has been reported as one of the most promising paths toward narrowing the band gap energy, and another benefit of N-doping is that it does not cause structural damage that could greatly reduce the photon conversion efficiency [12].

Testing the antibacterial responses of ATO nanotubes would provide valuable information for assessing the effects of annealing or N-doping of the nanotubular form of TiO₂ on its antibacterial activity. The motivation of this research was to investigate the photocatalytic antibacterial properties of ATO nanotubes derived by annealing at 450 °C or

550 °C, or by N-doping, under darkness and under illumination with UV-A or visible light.

2. Material and methods

2.1. The preparation of ATO nanotubes

To prepare the TiO₂ nanotubes, titanium foil (Aldrich, 99.7% purity, 0.127 mm thick) was immersed in electrolyte (0.5 wt% NH₄F ammonium fluoride mixed with (CH₂OH)₂ ethylene glycol) while clamped onto a holder before being anodized at 60 V for 1 h. Then the samples were annealed for 3 h at 450 °C in air, for 4 h at 550 °C in air, or for 4 h at 550 °C in a mixed atmosphere of N₂/Ar to form channel arrays of nanotubular anatase ATO. The bandgap energy of photoluminescence (PL) for the ATO nanotubular anatase was measured with a He–Cd laser at 10 K with a wavelength of 325–700 nm. The chemical elements of the N-doped samples were determined by X-ray photoelectron spectroscopy (XPS) with a resolution of 0.1 eV to check whether the nitrogen was doped inside the TiO₂. The excitation source was Al K_α monochromatic radiation (15 keV; 200 W). The O peak at 529 eV, the Ti peak at 459 eV, the N peaks respectively at 396 eV and 400 eV, and the C peak at 285 eV were analyzed at a resolution of 0.1 eV.

2.2. Antibacterial activity evaluation

E. coli was used to test the antibacterial effects of these ATO nanotubes. After inoculation of 0.1 ml of fresh *E. coli* culture into 10 ml of Nutrient Broth medium (BD Bioscience, USA) in test tubes containing 2 cm × 2 cm ATO nanotube layers, 1 cm × 1 cm ATO nanotube layers, or no ATO nanotubes, the cultures were incubated at 30 °C and 125 rpm under UV-A light (0.1 mW/cm²), visible light illumination, or dark conditions. The growth of *E. coli* in each culture was counted and determined by measuring its optical density (O.D.) at 600 nm spectrophotometrically every three hours. All experiments were performed in duplicate. For drawing the growth curves of these cultures, the average of the optical density at 600 nm were plotted against the time period.

3. Results and discussion

3.1. Physical properties of the different ATOs

The self-organized TiO₂ nanotube layers of anatase in the annealed and N-doped samples were observed by scanning electron microscopy. The tip view of the

N-doped ATO is shown in Fig. 1. The thickness of the nanotube layers was about 10 μm (inset). Similar top views of the 450 $^{\circ}\text{C}$ and 550 $^{\circ}\text{C}$ annealed ATO were observed respectively. The thickness of the nanotube layers was also about 10 μm .

The photoluminescence (PL) and XPS of the N-doped ATO nanotube layers are shown in Fig. 2. The conversion of the band gap energies of the peaks in the PL spectrum to the wavelengths is plotted in the inset of Fig. 2(a). The N-doped ATO nanotubes ($\text{TiO}_{2-x}\text{N}_x$) had a band gap energy (E_g) of about 2.47 eV, which corresponded to a wavelength of 500 nm. The 450 $^{\circ}\text{C}$ and 550 $^{\circ}\text{C}$ annealed ATO nanotubes had band gap energies of approximately 3.0 eV and 2.78 eV, respectively, which corresponded to wavelengths of 413 nm and 450 nm, respectively. Thus, these ATO nanotubes had a visible light photo response. The N-doped TiO_2 nanotube layer in Fig. 1 displayed the 1s core levels of the N spectrum shown in Fig. 2(b). Two peaks appeared at 396 ± 0.2 eV and 400 ± 0.2 eV for the nitrogen spectral signal (N 1s), respectively. The former was essentially atomic N or NO in the form of mixed titanium oxide-nitride ($\text{TiO}_{2-x}\text{N}_x$), which corresponded to the β -N state. The latter was molecularly chemisorbed N_2 , corresponding to the γ -N state. This indicates that the heat treatment of N-doping indeed successfully led to the substitution of nitrogen atoms at some oxygen sites [11,12].

3.2. Bacteriostasis of *E. coli* by ATO

The effects of the surface area of the ATO on photocatalytic antibacterial activity are shown under the dark condition in Fig. 3(a), UV-A light in Fig. 3(b), and visible light illumination in Fig. 3(c),

Fig. 1. SEM top views of the nanotube layers of ATO N-doped in a gas mixture of N_2/Ar . The inset shows the thickness of the nanotube layers in cross-section. Similar structures were observed in ATOs annealed at 450 $^{\circ}\text{C}$ and 550 $^{\circ}\text{C}$ respectively.

respectively. The N-doped and the 450 $^{\circ}\text{C}$ and 550 $^{\circ}\text{C}$ annealed ATO nanotubes had band gap energies corresponding to wavelengths in the visible range. Thus, these ATO nanotubes had a visible light photo response. These facts elucidated why these ATO nanotubes had antibacterial effects when illuminated under visible light but not under the UV-A light and dark conditions. As shown in Fig. 3(b), in the first 9 h, all ATO samples showed antibacterial effects; subsequently, however, most *E.*

Fig. 2. (a) PL and (b) XPS spectra of the N-doped ATO nanotubular layer shown in Fig. 1. The inset in (a) is a re-plotting of the data in wavelengths (nm).

Fig. 3. The effects of surface area and ATO treatment on the photocatalysis of TiO₂ were determined by measuring the growth of *E. coli* under (a) darkness (b) UV-A light illumination, and (c) visible light illumination.

coli cultures containing ATO samples started to grow faster than did those without ATO (labeled control in Fig. 3), but not faster than the *E. coli* cultures containing 2 cm × 2 cm layers of N-doped ATO. The growth of the cultures containing the 2 cm × 2 cm layers was inhibited more than that of the cultures containing 1 cm × 1 cm layers, whether they were annealed at 450 °C or 550 °C or were N-doped. As shown in Fig. 3(b), the 1 cm × 1 cm layer of ATO annealed at 550 °C had the weakest antibacterial effect, while the 2 cm × 2 cm layer of N-doped ATO had the strongest. The reason why most cultures containing ATO samples grew better than the control group after 12 h under UV-A illumination in Fig. 3(b) may be that the hydrophilic and rough surfaces of the nanotubes allowed the cultures to adhere to them, which enhanced the spread and proliferation of the cultures [13], leading to growth that outpaced the antibacterial activity under UV-A light, but not in cultures containing N-doped ATO with a larger area, the 2 cm × 2 cm layers. In contrast, the surface area of a small 1 cm × 1 cm layer of ATO was not large enough to generate sufficient ROS under UV-A light illumination.

On the other hand, under visible light illumination (Fig. 3(c)), all cultures containing ATO grew more slowly than did the cultures without ATO. Moreover, the growth of the cultures containing 2 cm × 2 cm layers of ATO was inhibited more than that of the cultures containing 1 cm × 1 cm layers, regardless of whether they were annealed at 450 °C or 550 °C or were N-doped. ATO annealed at 550 °C had the weakest antibacterial effect, while N-doped ATO had the strongest. The band gap energies of the ATO annealed at 450 °C, ATO annealed at 550 °C, and N-doped ATO were 3.0 eV, 2.78 eV and 2.47 eV, respectively. The antibacterial activity of the ATO annealed at 550 °C was weaker than that of the ATO annealed at 450 °C, possibly due to the coexistence of the anatase with a small amount of rutile. It has been reported previously that titanium oxide has better catalytic performance in the full anatase form than in the rutile form, than a mixture of the anatase and rutile forms, or than TiO₂ with an amorphous structure [14].

It has been reported that doping nitrogen into the TiO₂ structure to cause defects of oxygen vacancies can confer hydrophilicity to the treated specimens [15]. These vacancies can be occupied by water molecules to improve the wetting property of the surface, resulting in adsorbed hydroxyl groups [13], which contribute to the antibacterial activity under visible light illumination, especially in cultures

containing a larger surface area of both nanotubes annealed at 450 °C and N-doped ATO nanotubes.

4. Conclusions

Under dark conditions, all cultures containing ATO showed growth curves similar to those of the cultures without ATO. During the 24 h of incubation, the growth of *E. coli* cultures under UV-A illumination was much slower than that of cultures under visible light illumination, probably because the growth was affected by not only UV-A illumination but also the free radicals or superoxide ions produced by the ATO after UV-A illumination. Overall, ATO annealed at 450 °C, ATO annealed at 550 °C, and N-doped ATO had UV-A and visible light photo responses. N-doped ATO had the strongest photo response, and the larger surface area of the ATO had a positive effect on photo response. A larger surface area allows more generation of ROS, leading to a higher photocatalytic response. In addition, the hydrophilic surfaces of N-doped ATO tend to lead to a greater number of intermediates in oxidative formation. At the same time, during the photocatalytic process under light exposure, N-doped ATO promotes the generation of ROS necessary for bactericidal action. Therefore, the antibacterial activities of N-doped ATO nanotubes become more effective as the total surface area increases.

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